

This article was downloaded by:

On: 27 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Organic Preparations and Procedures International

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t902189982>

AN IMPROVED SYNTHESIS OF 2-HETEROARYL-3-PHENYL-4(3H)-QUINAZOLINONES

T. Hisano^a; K. Muraoka^a; M. Ichikawa^a

^a Faculty of Pharmaceutical Sciences, Kumamoto University, Kumamoto, JAPAN

To cite this Article Hisano, T. , Muraoka, K. and Ichikawa, M.(1977) 'AN IMPROVED SYNTHESIS OF 2-HETEROARYL-3-PHENYL-4(3H)-QUINAZOLINONES', *Organic Preparations and Procedures International*, 9: 1, 41 – 44

To link to this Article: DOI: 10.1080/00304947709355660

URL: <http://dx.doi.org/10.1080/00304947709355660>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

REFERENCES

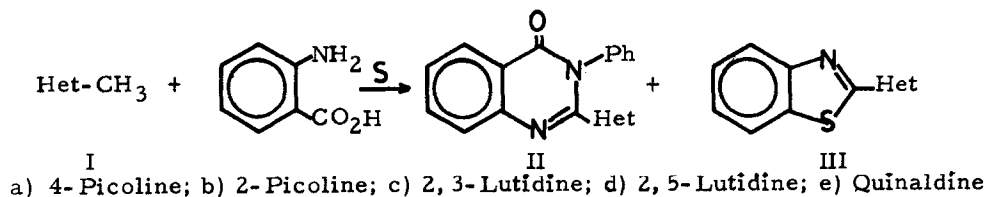
- * Present address: Brewing and Fermenting Dept., Allied Breweries (UK) Ltd., Burton-on-Trent, U.K.
1. D. O. Gray and L. F. Fowden, *Nature*, 189, 401 (1961).
 2. R. Walter, I. L. Schwartz, L. J. Traath, M. C. Berman and D. H. Schlessinger, *Can. J. Chem.*, 44, 2348 (1966).
 3. F. Weyland, P. Klinke and I. Eigen, *Chem. Ber.*, 90, 1896 (1957).
 4. Y. Liwshitz, A. Zilkha and Y. Anuea, *J. Am. Chem. Soc.*, 78, 3069 (1956).
 5. T. Hashizume, *J. Agr. Chem. Soc. Japan*, 25, 25 (1955).
 6. L. Fowden, *Biochem. J.*, 81, 154 (1961).
 7. R. W. Dineen and D. O. Gray, *J. Chromatog.*, 111, 248 (1975).

AN IMPROVED SYNTHESIS OF 2-HETEROARYL-3-PHENYL-4(3H)-QUINAZOLINONES

Submitted by T. Hisano*, K. Muraoka and M. Ichikawa
(1/3/77)

Faculty of Pharmaceutical Sciences
Kumamoto University
5-1 Oe-honmachi
Kumamoto 862
JAPAN

A modification of the Niementowski quinazolone synthesis¹ permits the obtention of 2-heteroaryl-3-aryl-4(3H)-quinazolinones by a one-step procedure in which the decarboxylation of a one molar excess of anthra-



OPPI BRIEFS

nilic acid under the high temperature conditions used, serves as the source of the aromatic amine in situ. Although the yields are poor to moderate, they are relatively good compared to those of previous preparations.² A by-product of this reaction is the 2-heteroaryl benzothiazoles (III),^{2b,3} formed from the reaction of I with aniline and sulfur.

EXPERIMENTAL

Most of products are listed in Table I. IR spectra were recorded on Nippon Bunko IR-G spectrophotometer and nmr spectra were taken with JNM-C-60H spectrometer in ca 5% (w/v) CHCl₃ solution with tetramethylsilane as an internal standard and chemical shifts were expressed in τ value.

Typical Procedure.- A mixture of 0.10 mole of 4-picoline (Ia), 0.20 mole of anthranilic acid and 0.30 mole of sulfur was heated in an oil bath at 195-200° for 8 hrs. After removal of unchanged 4-picoline in vacuo, the brown residue was dissolved in 80 ml. of benzene and kept overnight below 15°. The crystals which separated were collected by suction and recrystallized from EtOH, give 0.7 g. (1.6%) of the anthranilic acid salt of IIa, mp. 249°, as colorless prisms. IR ν_{\max}^{KBr} cm⁻¹: 3290 and 3340 (NH₂), 1686 (C=O), which was identical with 2-(4-pyridyl)-3-phenyl-4(3H)-quinazolinone anthranilate (mp. 249°)⁴ by the mixed melting point test.

Anal. Calcd. for C₁₉H₁₃ON₃·C₇H₇O₂N: C, 71.56; H, 4.58; N, 12.84.

Found: C, 71.63; H, 4.51; N, 12.56.

The mother liquor was concentrated to 40 ml. by distillation and then chromatographed over 180 g. of Al₂O₃ (200 mesh), benzene being used as eluent. From the first eluted fraction, a crude crystalline mass was obtained and recrystallized from pet. ether-benzene to give 11.9 g. (40%) of IIa, mp. 159°, as colorless prisms, which was identical with 2-(4-pyridyl)-3-phenyl-4(3H)-quinazolinone (mp. 159°)⁴ by mixture melting point

test. IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 1680 (C=O); nmr (in CDCl_3) at 60 Mc: τ 1.62 (2H, d-d, $J = 4.5$ Hz, pyridine C_2 - and C_6 -H, 1.76 [1H, d (broad), $J = 7.0$, 4(3H)-quinazolinone ring C_5 -H],^{5,6} 2.10-3.10 (10H, m, remaining aromatic C-H).

Anal. Calcd. for $\text{C}_{19}\text{H}_{13}\text{ON}_3$: C, 76.25; H, 4.35; N, 14.05; MW, 299.

Found: C, 76.31; H, 4.25; N, 13.75; MW (mass spectrum) 299.

From the second eluted fraction, a crude crystalline mass was obtained and recrystallized from pet. ether-benzene to give IIIa in 6% yield (1.2 g.), mp. 133-135°, as colorless plates, which was identical with 2-(4-pyridyl-benzthiazole (mp. 135°)³ by mixture melting point and by ir spectral comparison.

Table I. 2-Heteroaryl-3-phenyl-4(3H)-quinazolones^a

Cmpd	Appearance ^b	Heterocycle	mp. (°C)	Yield (%)	Analysis(%)		
					Calcd.	(Found)	
					C	H	N
IVa	colorless (prisms)	4-pyridyl	159 ⁴	40	76.25 (76.31)	4.35 4.25	14.05 13.75
IVb	colorless (amorphous)	2-pyridyl	162 ^{2b}	25	76.25 (76.23)	4.35 4.51	14.05 13.83
IVc	colorless prisms	2-(3-methyl-pyridyl)	159-161	15	76.66 (76.91)	4.83 4.57	13.41 13.36
IVd	colorless prisms	2-(5-methyl-pyridyl)	174	30	76.66 (76.47)	4.83 4.60	13.41 13.48
IVe	colorless needles ^c	2-quinolyl	186-187	30	79.08 (78.92)	4.30 4.41	12.03 11.98

a) All reactions were run using 1 equivalent of the heterocycle, 2 equivalents of anthranilic acid and 3 equivalents of sulfur. b) Recrystallized from pet ether/benzene except as otherwise noted c) From benzene.

OPPI BRIEFS

Anal. Calcd. for $C_{12}H_8N_2S$: C, 67.88; H, 3.81; N, 13.21; MW, 212.

Found: C, 68.14; H, 3.93; N, 12.92; MW (mass spectrum), 212.

Other compounds (IIb-3) were synthesized as described for IIa (Table I).

ACKNOWLEDGEMENT.- The authors wish to thank the members of the Analytical Department of this Faculty for the microanalyses and the spectral measurements. We are also grateful to Mr. M. Nakatomi, President of Hisamitsu Pharmaceutical Co. Inc. for the supply of several chemicals.

REFERENCES

1. T. Hisano. Org. Prep. Proced. Int., 5, 145 (1973).
2. a) T. Hisano and M. Ichikawa, Yakugaku Zasshi, 91, 727 (1971); b) T. Hisano and M. Ichikawa, Chem. Pharm. Bull. (Tokyo), 19, 2625 (1971).
c) T. Hisano, M. Ichikawa, G. Kito and T. Nishi, *ibid.*, 20, 2575 (1972). d) T. Hisano, M. Ichikawa, A. Nakagawa and M. Tsuji, *ibid.*, 23, 1910 (1975).
3. H. Saikachi and T. Hisano, *ibid.*, 7, 716 (1959).
4. T. Hisano, T. Nishi and M. Ichikawa, Yakugaku Zasshi, 92, 582 (1972).
5. R. H. Bible, Jr., "Guide to the NMR Empirical Method, A Work Book", Plenum Press, New York, 1967, p. 22.
6. T. Hisano, M. Ichikawa, K. Muraoka, Y. Yabuta, Y. Kido and M. Shibata, Chem. Pharm. Bull. (Tokyo), 24, 2244 (1976).